Federal Ministry for the Environment, Nature Conservation, Building and Nuclear Safety

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Importance of physical-chemical properties of ENM (intrinsic/extrinsic) for environmental hazard and risk assessment

Prof. Iseult Lynch University of Birmingham

Scientific Stakeholder meeting on nanomaterials in the Environment, 10 and 11 Oct 2017

Intrinsic versus extrinsic properties



Constant throughout life => Intrinsic to an individual Context dependent => changes with seasons, with exposure location etc.

Harder to imagine for NMs

Intrinsic properties are inherent to the nano-form of a material, and include e.g. structure and structural strain, shape, porosity, structural configuration and bandgap.

Extrinsic properties which are those connected to the surface area of the NM, including e.g. surface interactions and transformations of NM surface and biomolecules (e.g. unfolding, receptor activation, membrane damage, fibrillation etc.) as a result of binding.

Composition aspects such as inherent molecular toxicity, charge, hydrophobicity and coating (although also linked to both the intrinsic and extrinsic axes).

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Nano Today (2014) 9, 266-270

ECHA JRC RIVM approach

Intrinsic

"What they are?"

Chemical composition

Surface characteristics, including coating chemistry, functionalisation, surface charge

Particle size distribution (number average)

Specific surface area

Particle shape (e.g. aspect ratio)

Impurities

Crystalline phase(s) and size

Redox potential / band gap

Hydrophobicity/wettability

Rigidity

solubility

Isoelectric point

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Extrinsic

"Where they go, what they do?"

Rate of dissolution (in environment)

Dispersibility/agglomeration and dispersion stability

Dustiness (depends on moisture)

Biological (re)activity (or surface reactivity)

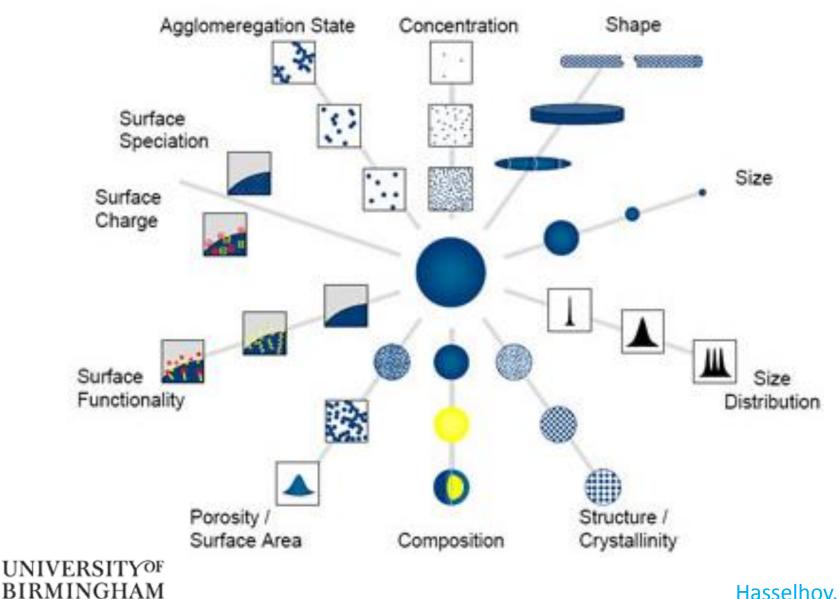
Photoreactivity

Zeta potential

Surface affinity

Discuss this tomorrow on Table 1 of Knowledge Cafe

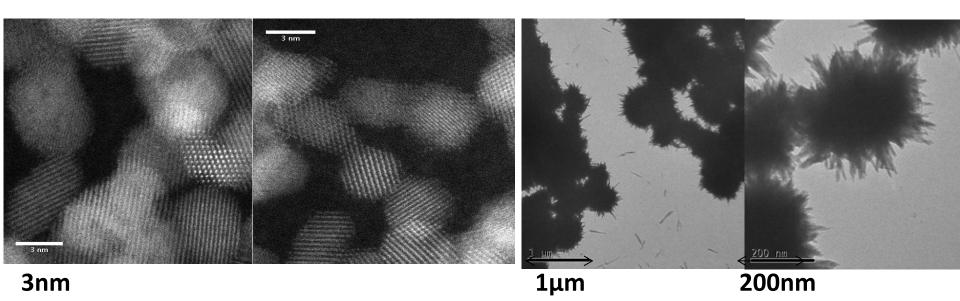
Dynamic nature of NMs in environment



Hasselhov, 2009

Parameter /	Context	Potential impacts of surroundings
Descriptor	dependent?	
Size /size distribution	Yes	In environment, most likely decreased by binding of NOM
		(stabilization). Protein binding may lead to either increased or
		decreased size via bridging or steric stabilisation.
		pH/ionic strength may alter agglomeration.
Surface area	Yes	Aggregation/agglomeration will reduce available surface area.
Purity (particle /	Maybe	Impurities / dispersants may be more effectively released from NM
dispersant)		surface under different environmental conditions.
Dissolution potential	Yes	pH, ionic strength, redox potential and adsorbed biomolecules
		affect dissolution rate.
Photochemical	Most likely	Differences in pH and ionic strength and presence/absence of
activity		organic matter may affect electron transfer and result in
		protonation of different excited states.
Surface charge /	Yes	Binding of ions/ biomolecules may confer a different charge /
chemistry		charge distribution and surface groups but this may be dynamic.
Hydrophobicity	Yes	Binding of biomolecules typically results in a more hydrophilic
		surface presentation, although may be dynamic.
Redox activity	Most likely	Different surfaces / coatings / bound ligands may result in different
		radical species being generated.(<u>Li, 2013</u>)
Shape	Most likely	Agglomeration will result in different overall shape. Bundling/
	\succ	unbundling of nanotubes is an example.
Crystal structure	Unlikely	Structure is a bulk property, established during the formation of an
		NM and cannot change by processes occurring on the surface,
		unless if the NM dissolves completely and re-precipitates.
Porosity / surface	Most likely	Dependent on pore size or nature of defect, most likely decreased
defects		due to biomolecule absorption; may be influenced by dissolution if
		NMs do not dissolve congruently or are a mixed phase.

NMs ageing in the environment



CeO₂ NMs + 5mM KH₂PO₄ + 5mM citric acid + 5mM ascorbic acid @ pH 5.5

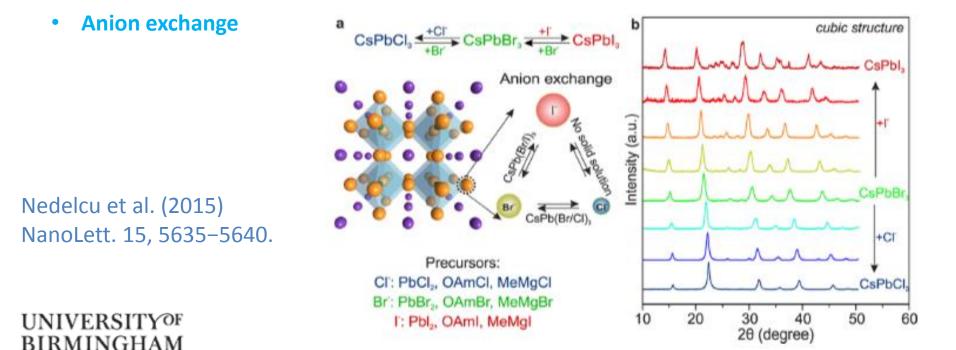
□ Characterisation by XRD, UV-vis, TEM



Perovskite materials

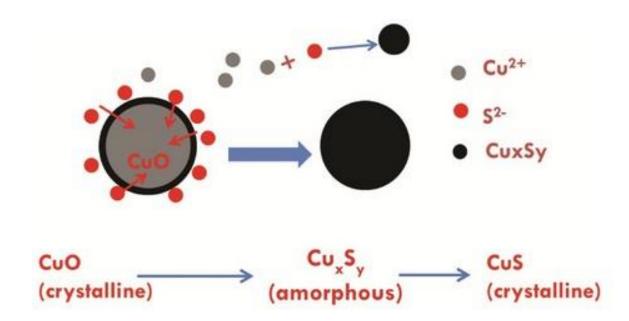
- Dissolution / Behaviour under reducing and non-reducing conditions
- Impact of light intensity on perovskites
- Acid–Base and Redox Properties

Thermodynamic metastability



Intrinsic = Context independent

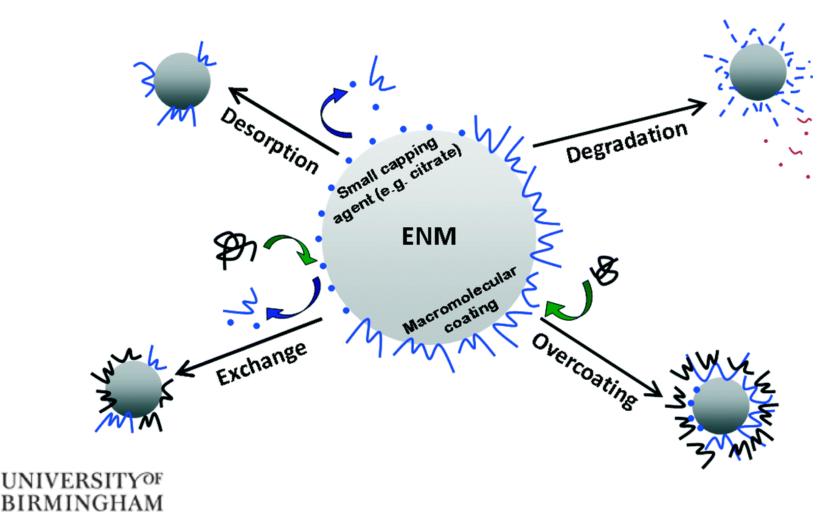
Are there any properties that are truly independent of context and unchanging over time?



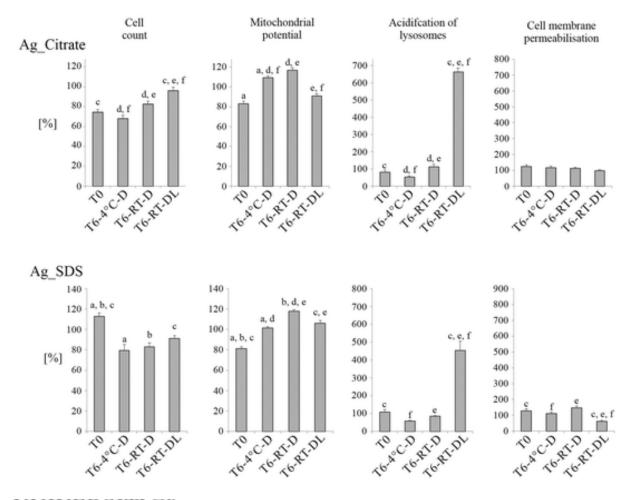
Should rate of change in defined (standardised) contexts be what we consider instead?

Extrinsic = Context dependent

Turns out nearly everything can be affected by the surroundings!



Ageing during storage an issue



Looked at the effect of storage conditions and coating structure on AgNP stability.

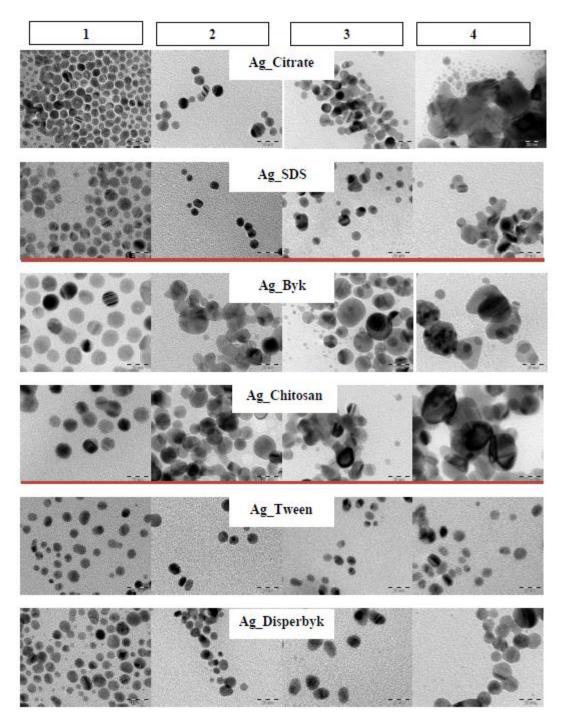
Conditions assessed include:

Room Temp – Light (RT-DL) Room Temp – Dark (RT-D) 4°C-Dark (4°C-D)

Times: 0, 3 and 6 months

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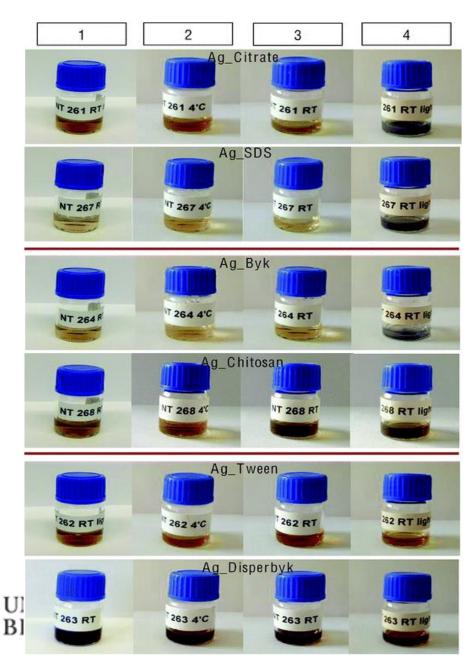
Izak-Neu et al., RSC Adv., 2015, 5, 84172



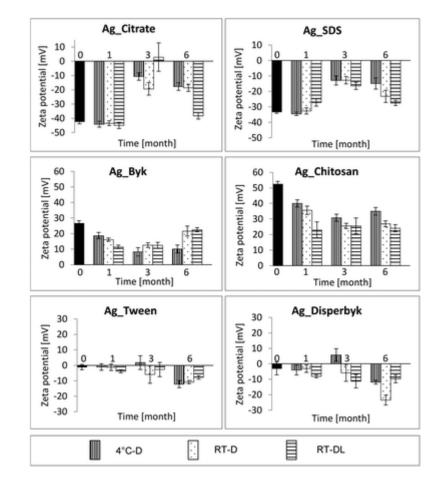
(1) T0,
 (2) T6:4 °C-D,
 (3) T6:RT-D,
 (4) T6:RT-DL
 (scale bar 20 nm).

Agglomeration
Dissolution
Loss of stabiliser
Surface chemistry evolution





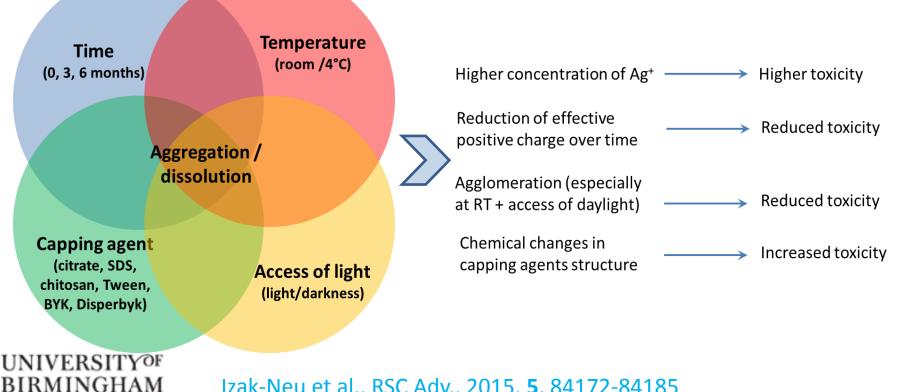
(1) T0, (2) T6:4 °C-D, (3) T6:RT-D; (4)T6:RT-DL



Zeta potential

Monitor "ageing" of NMs

Recommendation to add temporal assessment of NP properties from when NPs are synthesized / purchased / "opened" and during the experimental studies



Izak-Neu et al., RSC Adv., 2015, **5**, 84172-84185

Record of NM Provenance



Information about NM synthesis appropriate for sample provenance:

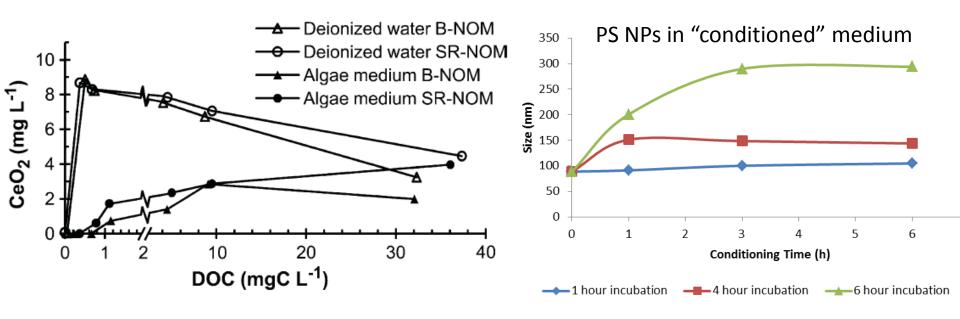
- (1) Record of sample synthesis: reference or details of synthesis as known (e.g., process, vendor, lot number, chemicals, and chemical sources)
- (2) Characterization results: data reports including relevant dates and processing of samples for analysis
- (3) Important dates and times: synthesis, arrival in laboratory, opening of sample container, primary analysis measurements, and expiry date
- (4) Storage time, conditions, and containers: temperature, humidity, media, light shielded, shipping, or transport)
- (5) Record of additional processing: e.g., dried, washed, heated, sonicated, functionalized (including the method and number of times processed).

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Baer, Biointerphases. 2016 11(4): 04B401.

Characterisation in relevant media over relevant timeframes

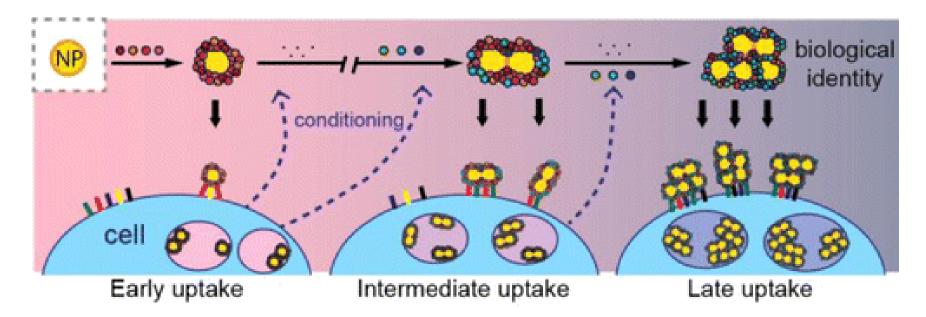
Dispersion medium typically determines agglomeration degree and affect dosimetry!



Quik et al., Chemosphere, 2010, 81: 711-715

Nasser et al, J Proteomics, 2016

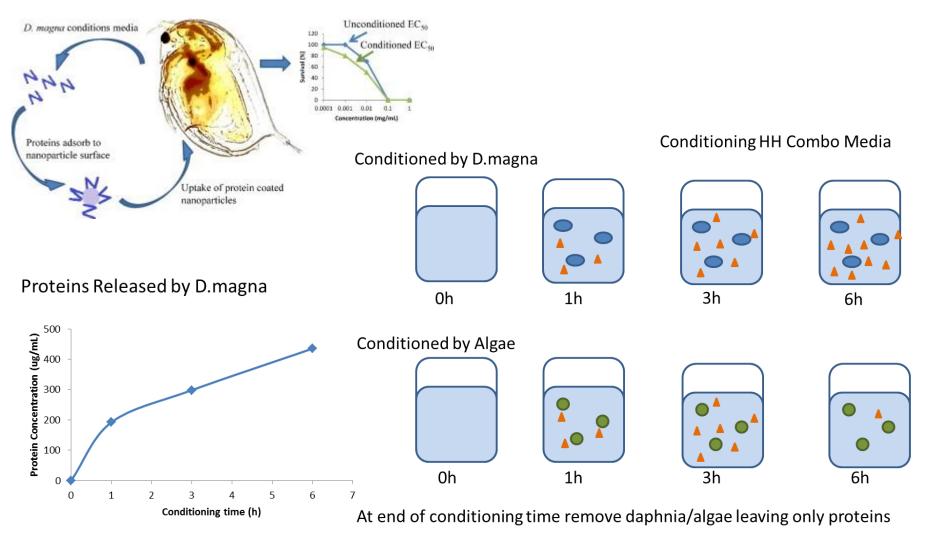
Dynamic nature of the system e.g. Cells / organisms "condition" media => evolution of form over time



Cells respond to the presence / uptake of NMs, leading to secretion of proteins and other biomolecules that cause the "initial" corona to evolve and which can lead to altered uptake and impacts. Not currently considered in assessing toxicity.

UNIVERSITY^{OF} BIRMINGHAM Albanese and Chan, *ACS Nano*, 2014, 8: 5515–5526 **DOI:** 10.1021/nn4061012

Secreted biomoelcule corona





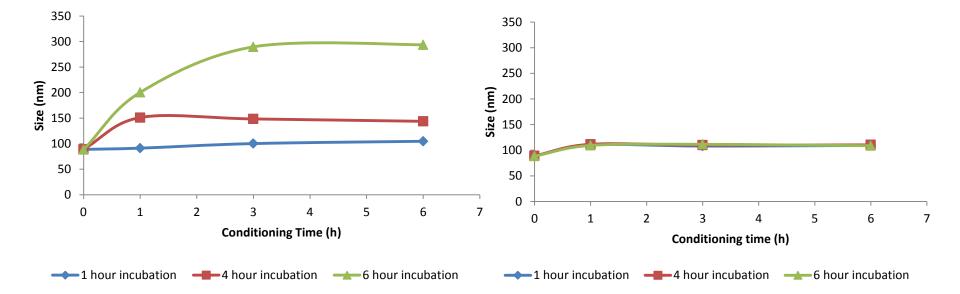
Work from PhD student Fatima Nasser

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Secreted biomolecule corona

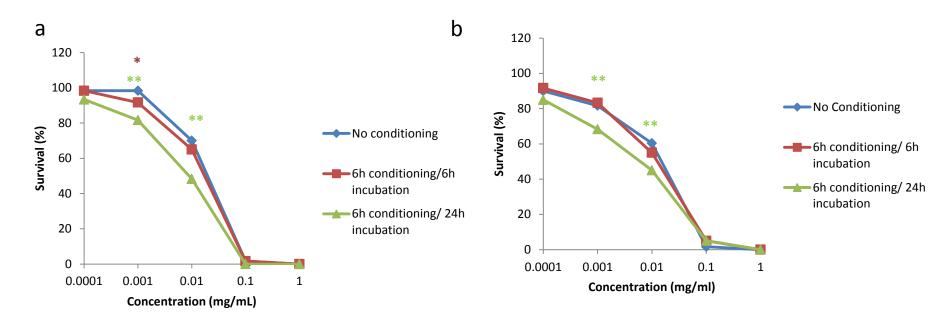
Incubation of Particles in Conditioned Media and Particle stability

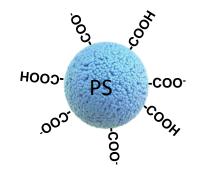


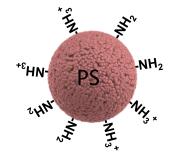
- Longer incubation time of NPs in conditioned media increases NP-protein interaction
- As conditioning time increases (as well as incubation time), agglomeration also increases indicating proteins acting as a potential destabilizer

UNIVERSITY^{OF} BIRMINGHAM • Low dose of proteins present in media causes no significant changes in NP stability

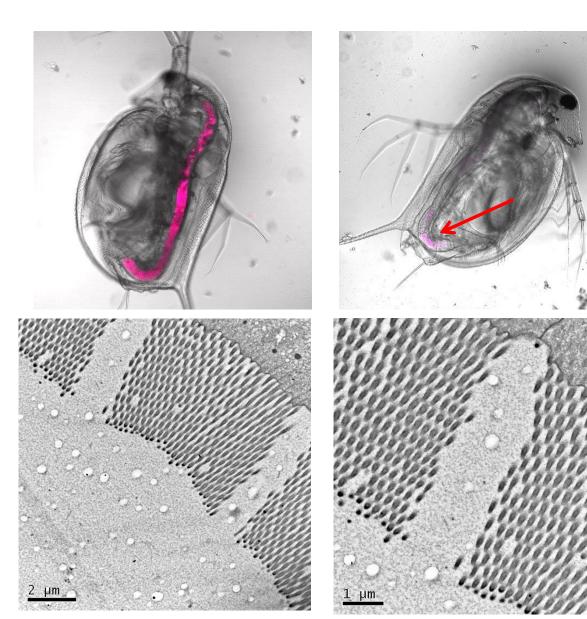
Eco-corona impacts EC₅₀







NM Retention in gut – impacts?



No evidence of translocation of PS NPs

Size matters – 500nm taken up more than 50nm NPs - Smaller NPs retained more!

Shape matters

Conditioning matters

Acknowledgements









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FP7 & H2020 project partners and many more...